



CORE LABORATORIES (U.K.) LIMITED

Advanced Technology Centre

SOLUBILITY-SWELLING STUDY

for

Statoil a.s.

15/9-19A

RFLA 980053

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Advanced Technology Centre

23rd November 1998

Statoil a.s.
Postboks 300,
Forus N4001,
Stavanger
Norway

Attention : Tore Tjomsland

Dear Sirs,

Subject : Solubility-swelling study; well 15/9-19A; our file RFLA 980053.

Core Laboratories (U.K.) Limited have recently completed a solubility-swelling study on samples from well 15/9-19A and are pleased to present the results of these analyses in the following report.

Part of this study was to prepare and despatch a large volume recombined sample for P.R.I. Canada. This sample was despatched on 26th June 1998.

The sample cylinders will remain at Core Laboratories (U.K.) Limited until otherwise instructed.

Core Laboratories are very pleased to have been of service to Statoil a.s. in this work. Should any questions arise concerning the data presented in this report, or if Core Laboratories may be of assistance in any other matter, please do not hesitate to contact us.

Yours sincerely,

Core Laboratories (U.K.) Limited
Reservoir Fluid Analysis

Neil Tosh
PVT Laboratory Supervisor





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Section A
Summary of PVT Methods and Data



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Summary of PVT Analysis Methods

Quality Checks

The saturation pressure of the separator liquid sample was attempted at room temperature as a quality check. However, the sample was found to be in two phases at the maximum working pressure of the cylinder upon opening. The separator liquid sample was heated to approximately 93°C prior to sample removal to eliminate wax precipitation and transferred to a large volume high pressure cylinder. The bubble point was then measured at 32.1 bara which suggested that gas had been lost due to opening the cylinder whilst the sample was in two phases.

A second separator liquid sample was delivered and the same scenario was encountered. On this occasion the cylinder was not opened which prevented any loss of gas. However, the sample volume was too large to perform a total fluid transfer to a high pressure cylinder. After discussion with Statoil a.s. it was agreed that a gas rich portion of the separator sample be transferred and then used with the separator gas sample to recombine to a specified bubble point.

The separator gas sample was heated to a temperature 20°F above separator temperature, agitated, and allowed to stabilise. The opening pressure, air content and hydrocarbon composition were then determined.

Liquid Composition

The separator liquid composition was determined using a flash/separation technique. The liquid was flashed at 120°F to obtain gas and liquid components. The gas was analysed using the extended analysis technique (see below) and the liquid by temperature programmed chromatography. The two analyses were then combined mathematically to the flash gas/liquid ratio.

Gas Data

Heptanes plus gas analysis was determined by routine gas chromatography using the GPA 2261-90 method. Extended gas analysis, to decanes plus, was determined by routine and extended chromatography. The extended chromatography was temperature programmed to improve the detection and resolution of the heptanes plus components. The two analyses were then combined mathematically.

The gas compressibility factor Z was calculated at 552.6 bara and the range of laboratory temperatures prevailing during the PVT study, using the method of Dranchuk, Purvis and Robinson, Institute of Petroleum Technical Series, No IP 74-008, 1974. From these values of Z the gas densities were also calculated for each gas addition and hence the weights of gas added.

Physical recombination

A measured volume of the gas rich separator liquid was charged to a high pressure visual cell at reservoir temperature (110°C). A known quantity of separator gas was added to the cell and the cell contents were stabilised in single phase at 552.6 bara. The saturation pressure of this recombined fluid was then determined. A second gas addition was performed to produce a recombined fluid which would have a saturation pressure close to the target value of 235.5 bara at 110°C. A partial constant mass expansion was performed and the bubble point was confirmed at 234.3 bara.



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Summary of PVT Analysis Methods

Physical recombination (continued)

The recombined sample composition was determined, as detailed earlier, and the composition of the gas deficient separator liquid was also determined. A theoretical recombination ratio was used in conjunction with the compositions of the separator gas and gas deficient separator liquid samples to calculate a mathematical wellstream composition that matched the measured recombined fluid composition. The samples were then physically recombined to this ratio and the composition of the recombined fluid was then measured and agreed closely with the calculated wellstream composition.

Solubility-Swelling Study

A portion of the recombined fluid was charged to a high pressure visual cell at 552.6 bara and thermally expanded to the reservoir temperature of 110°C. A constant composition expansion was carried out and the bubble point pressure was determined. Pressure-volume data for the single phase liquid were also measured. The density was determined at 552.6 bara by pumping a measured volume from the PVT cell for direct weighing. This was repeated until data consistent to within 1% were obtained. Density data for the single phase liquid were then derived from the volumetric data.

A measured portion of injection gas was added to the cell at 552.6 bara. The cell contents were stabilised in single phase at this pressure and the change in volume due to the added gas was noted. The density of each mixture was calculated from the known weights of the mixture components and the measured volume. A constant composition expansion was then carried out from 552.6 bara during which the saturation pressure was determined. Pressure-volume data for the single phase and two phase fluid and the liquid volumes in the two phase region were also measured. This was repeated for three further gas additions.

Compositional Calculations for the Solubility-Swelling Mixtures

The gas-oil ratio values for each mixture were used together with the compositions of the initial recombined fluid and injection gas to calculate the mixture compositions.



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SUMMARY OF SOLUBILITY-SWELLING DATA INJECTION GAS = SEPARATOR GAS, CYLINDER 50249

Solubility-Swelling Study at 110°C				
Moles of Gas per 1.000 Mole of Reservoir Fluid	Gas-Oil Ratio (Sm ³ /m ³) (1)	Vol. at Saturation Pressure Relative to Original Fluid	Bubble point Pressure (bara)	Density at Sat. Press. (g cm ⁻³)
0.000	0	1.0000	234.3	0.7350
0.114	16.5	1.0389	278.9	0.7217
0.439	63.7	1.1515	391.3	0.6876
0.683	99.1	1.2376	459.5	0.6653
1.024	148.7	1.3572	536.3	0.6392

SUMMARY OF SOLUBILITY-SWELLING DATA INJECTION GAS = CARBON DIOXIDE

Solubility-Swelling Study at 110°C					
Moles of Gas per 1.000 Mole of Reservoir Fluid	Gas-Oil Ratio (Sm ³ /m ³) (1)	Vol. at Saturation Pressure Relative to Original Fluid	Saturation Pressure (bara)		Density at Sat. Press. (g cm ⁻³)
0.000	0	1.0000	234.3	bp	0.7350
0.114	16.5	1.0413	257.2	bp	0.7353
0.438	63.7	1.1549	314.0	bp	0.7391
1.032	149.8	1.3551	402.3	bp	0.7483
2.324	337.5	1.7720	535.4	dp	0.7694

Note: bp - bubble point, dp - dew point.

(1) Cubic meters of gas at 1.0132 bara and 15.0°C per cubic meter of saturated recombined fluid at 234.3 bara and 110°C.



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Section B
Samples Received and Compositional Analysis



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SUMMARY OF SAMPLES RECEIVED

SEPARATOR GASES							
Sample Number	Cylinder Number	Sampling :-		Laboratory opening :-		Liquid Hydro-C. (cm ³)	Air Content (mole %)
		Pressure (bara)	Temp. (°C)	Pressure (bara)	Temp. (°C)		
93	50249	47.7	58.5	56.2	68.3	0	0.03

SEPARATOR LIQUIDS							
Sample Number	Cylinder Number	Sampling :-		Laboratory Bubble point :-		Water Recovered (cm ³)	Sample Volume (cm ³)
		Pressure (bara)	Temp. (°C)	Pressure (bara)	Temp. (°C)		
97	50273	47.7	58.5	Not possible		0	18000
100	55015	47.7	58.5	Not possible		0	2000

Note

Due to cylinder pressure limitations, neither separator liquid sample was able to be quality checked by determining ambient temperature bubble point pressure. Both samples existed in two phases at maximum cylinder working pressure.



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MEASURED COMPOSITION OF SEPARATOR GAS SAMPLE - CYLINDER 50249

Component	Mole %
Hydrogen	0.00
Hydrogen sulphide	0.00
Carbon dioxide	2.94
Nitrogen	1.15
Methane	80.54
Ethane	7.86
Propane	4.50
i-Butane	0.42
n-Butane	1.21
neo-Pentane	0.00
i-Pentane	0.27
n-Pentane	0.36
Hexanes	0.28
Me-Cyclo-pentane	0.06
Benzene	0.04
Cyclo-hexane	0.05
Heptanes	0.11
Me-Cyclo-hexane	0.03
Toluene	0.03
Octanes	0.06
Ethyl-benzene	0.00
Meta/Para-xylene	0.00
Ortho-xylene	0.00
Nonanes	0.03
Decanes plus	0.06
Totals	100.00
Calculated properties	
Heptanes plus properties	
Mole %	0.47
Molecular Weight (g mol ⁻¹)	102.0
Density at 15°C (g cm ⁻³)	0.7570
Real Relative Density (Air = 1 at 1.0132 bara and 15°C)	
	0.728

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HYDROCARBON COMPOSITION OF RECOMBINED FLUID
(Recombined to bubble point = 235.5 Bara)

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	1.74	0.64
Nitrogen	0.56	0.13
Methane	41.88	5.62
Ethane	6.26	1.57
Propane	5.70	2.10
i-Butane	0.80	0.39
n-Butane	2.91	1.41
neo-Pentane	0.00	0.00
i-Pentane	1.06	0.64
n-Pentane	1.69	1.02
Hexanes	2.22	1.60
Me-Cyclo-pentane	0.67	0.47
Benzene	0.58	0.38
Cyclo-hexane	0.46	0.32
Heptanes	1.81	1.51
Me-Cyclo-hexane	0.57	0.47
Toluene	0.72	0.56
Octanes	2.03	1.94
Ethyl-benzene	0.36	0.32
Meta/Para-xylene	0.32	0.28
Ortho-xylene	0.19	0.17
Nonanes	1.66	1.78
Tri-Me-benzene	0.10	0.10
Decanes	2.00	2.38
Undecanes	1.89	2.33
Dodecanes	1.68	2.26
Tridecanes	1.68	2.46
Tetradecanes	1.44	2.29
Pentadecanes	1.44	2.48
Hexadecanes	1.15	2.13
Heptadecanes	1.06	2.09
Octadecanes	1.04	2.17
Nonadecanes	1.05	2.31
Eicosanes	0.75	1.72
Heneicosanes	0.68	1.66
Docosanes	0.64	1.63
Tricosanes	0.57	1.51
Tetracosanes	0.54	1.50
Pentacosanes	0.49	1.40
Hexacosanes	0.48	1.44
Heptacosanes	0.43	1.34
Octacosanes	0.42	1.36
Nonacosanes	0.39	1.32
Triacontanes	0.40	1.39

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COMPOSITION RESIDUE PROPERTIES
(Recombined to bubble point = 235.5 Bara)

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.39	1.39		
Dotriacontanes	0.33	1.24		
Tritriacontanes	0.32	1.21		
Tettratriacontanes	0.30	1.17		
Pentatriacontanes	0.29	1.18		
Hexatriacontanes plus	3.86	31.22		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm ⁻³)	Molecular weight (g mol ⁻¹)
Heptanes	3.52	2.68	0.7322	91.4
Octanes	3.32	2.96	0.7432	106.7
Nonanes	2.53	2.54	0.7613	120.7
Decanes	2.10	2.48	0.7384	141.2
Calculated residue properties				
Heptanes plus				
Mole %		35.18		
Molecular Weight (g mol ⁻¹)		289		
Density at 15.6°C (g cm ⁻³)		0.9079		
Decanes plus				
Mole %		25.81		
Molecular Weight (g mol ⁻¹)		356		
Density at 15.6°C (g cm ⁻³)		0.9296		
Eicosanes plus				
Mole %		11.28		
Molecular Weight (g mol ⁻¹)		570		
Density at 15.6°C (g cm ⁻³)		0.9885		
Hexatriacontanes plus				
Mole %		3.86		
Molecular Weight (g mol ⁻¹)		968		
Density at 15.6°C (g cm ⁻³)		1.0736		
Whole sample properties				
Average mole weight (g mol ⁻¹)		120		



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Section C
Hydrocarbon Gas Solubility Swelling



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**CONSTANT COMPOSITION EXPANSION AT 110°C
RECOMBINED SAMPLE**

Pressure (bara)		Relative Volume (1)	Liquid Volume % (2)	Density (g.cm-3)	Y-function (3)
552.6		0.9570		0.7680	
414.7		0.9727		0.7557	
345.8		0.9820		0.7485	
276.8		0.9926		0.7405	
269.9		0.9938		0.7396	
263.0		0.9949		0.7387	
256.1		0.9961		0.7379	
249.2		0.9973		0.7370	
242.3		0.9986		0.7361	
235.4		0.9998		0.7351	
234.3	Bubble point pressure	1.0000	100.00	0.7350	
233.7		1.0006	99.82		
232.5		1.0020	99.51		
231.4		1.0031	99.29		
230.5		1.0041	99.10		
229.4		1.0053	98.91		
228.4		1.0064	98.74		
216.0		1.0213	97.01		
196.3		1.0506	94.81		3.831
168.8		1.1074	92.60		3.619
140.4		1.1971			3.394
117.0		1.3133			3.201
97.5		1.4624			3.036
81.8		1.6428			2.900
68.7		1.8677			2.781
59.2		2.0980			2.694
51.6		2.3499			2.622
39.7		2.9577			2.505
30.2		3.8025			2.408
21.8		5.2175			2.316

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
(2) Liquid shrinkage as a percentage of volume at bubble point pressure.
(3) Y-function = $(P_{sat} - P) / (P_{abs})(V/V_{sat} - 1)$



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CONSTANT COMPOSITION EXPANSION AT 110°C 1st GAS ADDITION - 0.114 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)	Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)	
552.6	0.9597	0.9970		0.7520	
483.6	0.9679	1.0055		0.7456	
414.7	0.9772	1.0152		0.7385	
345.8	0.9880	1.0264		0.7305	
318.2	0.9927	1.0313		0.7270	
311.3	0.9939	1.0326		0.7261	
304.4	0.9952	1.0339		0.7251	
297.5	0.9965	1.0352		0.7242	
290.6	0.9978	1.0366		0.7233	
283.7	0.9991	1.0379		0.7223	
278.9	Bubble point pressure	1.0000	1.0389	100.00	0.7217
276.3	1.0020	1.0410	99.52		
274.3	1.0036	1.0427	99.18		
272.9	1.0048	1.0439	98.94		
271.4	1.0061	1.0452	98.70		
269.8	1.0074	1.0466	98.47		
268.5	1.0085	1.0477	98.28		
265.8	1.0110	1.0503	97.90		
257.9	1.0183	1.0579	96.96		
244.3	1.0327	1.0728	95.58		
222.3	1.0613	1.1026	93.64		
191.6	1.1167	1.1601	91.01		
160.8	1.2011	1.2478			
134.6	1.3129	1.3639			
112.9	1.4547	1.5113			
95.3	1.6275	1.6908			
81.0	1.8318	1.9030			
70.3	2.0466	2.1262			
63.8	2.2181	2.3044			
48.5	2.8154	2.9249			
37.0	3.6221	3.7629			
31.8	4.1950	4.3582			
26.5	5.0183	5.2135			

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
 (2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
 (3) Liquid shrinkage as a percentage of volume at bubble point pressure.



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CALCULATED COMPOSITION OF MIXTURE FROM THE 1st HYDROCARBON GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	1.86	0.75
Nitrogen	0.62	0.16
Methane	45.75	6.68
Ethane	6.42	1.76
Propane	5.58	2.24
i-Butane	0.76	0.40
n-Butane	2.74	1.45
neo-Pentane	0.00	0.00
i-Pentane	0.98	0.64
n-Pentane	1.56	1.02
Hexanes	2.03	1.59
Me-Cyclo-pentane	0.61	0.47
Benzene	0.53	0.37
Cyclo-hexane	0.42	0.32
Heptanes	1.64	1.50
Me-Cyclo-hexane	0.52	0.46
Toluene	0.65	0.55
Octanes	1.83	1.91
Ethyl-benzene	0.32	0.31
Meta/Para-xylene	0.29	0.28
Ortho-xylene	0.17	0.17
Nonanes	1.50	1.75
Tri-Me-benzene	0.09	0.10
Decanes	1.80	2.34
Undecanes	1.70	2.28
Dodecanes	1.51	2.22
Tridecanes	1.51	2.41
Tetradecanes	1.30	2.24
Pentadecanes	1.30	2.43
Hexadecanes	1.03	2.09
Heptadecanes	0.95	2.06
Octadecanes	0.94	2.14
Nonadecanes	0.94	2.26
Eicosanes	0.67	1.69
Heneicosanes	0.61	1.62
Docosanes	0.58	1.60
Tricosanes	0.51	1.49
Tetracosanes	0.49	1.46
Pentacosanes	0.44	1.39
Hexacosanes	0.43	1.41
Heptacosanes	0.39	1.32
Octacosanes	0.38	1.34
Nonacosanes	0.35	1.28
Triacotanes	0.36	1.36



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COMPOSITION RESIDUE PROPERTIES FROM THE 1st HYDROCARBON GAS ADDITION

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.35	1.37		
Dotriacontanes	0.30	1.20		
Tritriacontanes	0.29	1.20		
Tettriacontanes	0.27	1.16		
Pentatriacontanes	0.26	1.15		
Hexatriacontanes plus	3.47	30.61		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm ⁻³)	Molecular weight (g mol ⁻¹)
Heptanes	3.19	2.66	0.7319	91.4
Octanes	3.00	2.92	0.7432	106.7
Nonanes	2.28	2.51	0.7614	120.7
Decanes	1.89	2.44	0.7384	141.2
Calculated residue properties				
Heptanes plus				
Mole %		31.71		
Molecular Weight (g mol ⁻¹)		289		
Density at 15.6°C (g cm ⁻³)		0.9077		
Decanes plus				
Mole %		23.23		
Molecular Weight (g mol ⁻¹)		356		
Density at 15.6°C (g cm ⁻³)		0.9296		
Eicosanes plus				
Mole %		10.15		
Molecular Weight (g mol ⁻¹)		570		
Density at 15.6°C (g cm ⁻³)		0.9885		
Hexatriacontanes plus				
Mole %		3.47		
Molecular Weight (g mol ⁻¹)		968		
Density at 15.6°C (g cm ⁻³)		1.0736		
Whole sample properties				
Average mole weight (g mol ⁻¹)		109.6		

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CONSTANT COMPOSITION EXPANSION AT 110°C
2nd GAS ADDITION - 0.439 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)		Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)
552.6		0.9710	1.1181		0.7081
483.6		0.9819	1.1307		0.7003
435.4		0.9907	1.1407		0.6941
428.5		0.9920	1.1423		0.6931
421.6		0.9934	1.1439		0.6922
414.7		0.9948	1.1455		0.6912
407.8		0.9963	1.1472		0.6902
400.9		0.9978	1.1489		0.6891
394.0		0.9994	1.1507		0.6881
391.3	Bubble point pressure	1.0000	1.1515	100.00	0.6876
388.2		1.0014	1.1531	98.75	
386.1		1.0024	1.1542	98.11	
383.9		1.0034	1.1554	97.57	
381.9		1.0044	1.1565	97.13	
379.9		1.0054	1.1577	96.74	
376.1		1.0073	1.1599	96.12	
354.3		1.0195	1.1740	93.39	
320.2		1.0438	1.2020	89.63	
271.9		1.0941	1.2599		
226.4		1.1702	1.3475		
188.6		1.2720	1.4647		
158.4		1.3988	1.6107		
134.2		1.5520	1.7871		
117.1		1.7054	1.9638		
104.2		1.8607	2.1425		
96.5		1.9763	2.2757		
72.8		2.5062	2.8858		
54.6		3.2605	3.7544		
38.8		4.5342	5.2210		

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
- (2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
- (3) Liquid shrinkage as a percentage of volume at bubble point pressure.



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CALCULATED COMPOSITION OF MIXTURE FROM THE 2nd HYDROCARBON GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	2.10	1.03
Nitrogen	0.74	0.23
Methane	53.48	9.52
Ethane	6.74	2.25
Propane	5.34	2.61
i-Butane	0.69	0.44
n-Butane	2.40	1.55
neo-Pentane	0.00	0.00
i-Pentane	0.82	0.66
n-Pentane	1.29	1.03
Hexanes	1.64	1.57
Me-Cyclo-pentane	0.49	0.45
Benzene	0.42	0.36
Cyclo-hexane	0.34	0.31
Heptanes	1.30	1.45
Me-Cyclo-hexane	0.41	0.44
Toluene	0.51	0.52
Octanes	1.44	1.82
Ethyl-benzene	0.25	0.30
Meta/Para-xylene	0.22	0.26
Ortho-xylene	0.13	0.16
Nonanes	1.17	1.67
Tri-Me-benzene	0.07	0.09
Decanes	1.41	2.23
Undecanes	1.33	2.17
Dodecanes	1.18	2.11
Tridecanes	1.18	2.28
Tetradecanes	1.01	2.13
Pentadecanes	1.01	2.30
Hexadecanes	0.81	1.98
Heptadecanes	0.74	1.95
Octadecanes	0.73	2.03
Nonadecanes	0.74	2.15
Eicosanes	0.53	1.60
Heneicosanes	0.48	1.54
Docosanes	0.45	1.52
Tricosanes	0.40	1.41
Tetracosanes	0.38	1.39
Pentacosanes	0.34	1.31
Hexacosanes	0.34	1.34
Heptacosanes	0.30	1.25
Octacosanes	0.29	1.27
Nonacosanes	0.27	1.22
Triacotanes	0.28	1.29

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**COMPOSITION RESIDUE PROPERTIES
 FROM THE 2nd HYDROCARBON GAS ADDITION**

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.27	1.30		
Dotriacontanes	0.23	1.14		
Triatriacontanes	0.22	1.14		
Tetraatriacontanes	0.21	1.10		
Pentatriacontanes	0.20	1.10		
Hexatriacontanes plus	2.70	29.03		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm ⁻³)	Molecular weight (g mol ⁻¹)
Heptanes	2.54	2.58	0.7323	91.4
Octanes	2.36	2.79	0.7432	106.7
Nonanes	1.78	2.39	0.7611	120.7
Decanes	1.48	2.32	0.7383	141.2
Calculated residue properties				
Heptanes plus				
Mole %	24.77			
Molecular Weight (g mol ⁻¹)	288			
Density at 15.6°C (g cm ⁻³)	0.9074			
Decanes plus				
Mole %	18.09			
Molecular Weight (g mol ⁻¹)	355			
Density at 15.6°C (g cm ⁻³)	0.9295			
Eicosanes plus				
Mole %	7.90			
Molecular Weight (g mol ⁻¹)	570			
Density at 15.6°C (g cm ⁻³)	0.9885			
Hexatriacontanes plus				
Mole %	2.70			
Molecular Weight (g mol ⁻¹)	968			
Density at 15.6°C (g cm ⁻³)	1.0736			
Whole sample properties				
Average mole weight (g mol ⁻¹)	90.1			

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CONSTANT COMPOSITION EXPANSION AT 110°C
3rd GAS ADDITION - 0.683 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)	Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)	
552.6	0.9805	1.2134		0.6786	
518.1	0.9872	1.2218		0.6739	
490.5	0.9930	1.2290		0.6700	
483.6	0.9945	1.2308		0.6690	
476.8	0.9961	1.2327		0.6679	
469.9	0.9976	1.2347		0.6669	
463.0	0.9992	1.2366		0.6658	
459.5	Bubble point pressure	1.0000	1.2376	100.00	0.6653
454.5	1.0017	1.2397	97.31		
451.0	1.0030	1.2413	95.94		
447.8	1.0042	1.2428	94.94		
444.5	1.0054	1.2443	94.07		
441.7	1.0065	1.2456	93.45		
437.1	1.0083	1.2479	92.52		
413.3	1.0188	1.2609	88.88		
374.5	1.0405	1.2877	84.61		
318.5	1.0868	1.3450			
265.4	1.1580	1.4331			
221.9	1.2525	1.5501			
186.6	1.3726	1.6988			
158.8	1.5154	1.8754			
133.6	1.7070	2.1125			
112.4	1.9465	2.4089			
102.9	2.0907	2.5875			
80.6	2.5796	3.1924			
62.5	3.2552	4.0287			
54.0	3.7386	4.6269			
45.3	4.4303	5.4829			

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
(2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
(3) Liquid shrinkage as a percentage of volume at bubble point pressure.

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**CALCULATED COMPOSITION OF MIXTURE
 FROM THE 3rd HYDROCARBON GAS ADDITION**

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	2.22	1.22
Nitrogen	0.80	0.28
Methane	57.34	11.47
Ethane	6.90	2.59
Propane	5.22	2.87
i-Butane	0.65	0.47
n-Butane	2.23	1.62
neo-Pentane	0.00	0.00
i-Pentane	0.74	0.67
n-Pentane	1.16	1.04
Hexanes	1.44	1.55
Me-Cyclo-pentane	0.43	0.45
Benzene	0.36	0.35
Cyclo-hexane	0.30	0.31
Heptanes	1.13	1.41
Me-Cyclo-hexane	0.35	0.43
Toluene	0.44	0.51
Octanes	1.24	1.77
Ethyl-benzene	0.22	0.29
Meta/Para-xylene	0.19	0.25
Ortho-xylene	0.11	0.15
Nonanes	1.01	1.61
Tri-Me-benzene	0.06	0.09
Decanes	1.21	2.15
Undecanes	1.14	2.09
Dodecanes	1.01	2.03
Tridecanes	1.01	2.20
Tetradecanes	0.86	2.05
Pentadecanes	0.86	2.22
Hexadecanes	0.69	1.91
Heptadecanes	0.64	1.88
Octadecanes	0.62	1.95
Nonadecanes	0.63	2.07
Eicosanes	0.45	1.54
Heneicosanes	0.41	1.48
Docosanes	0.38	1.46
Tricosanes	0.34	1.36
Tetracosanes	0.32	1.34
Pentacosanes	0.29	1.26
Hexacosanes	0.29	1.29
Heptacosanes	0.26	1.20
Octacosanes	0.25	1.22
Nonacosanes	0.23	1.17
Triacotanes	0.24	1.24

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**COMPOSITION RESIDUE PROPERTIES
 FROM THE 3rd HYDROCARBON GAS ADDITION**

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.23	1.25		
Dotriacontanes	0.20	1.10		
Tritriacontanes	0.19	1.10		
Tetratriacontanes	0.18	1.06		
Pentatriacontanes	0.17	1.05		
Hexatriacontanes plus	2.32	27.95		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm-3)	Molecular weight (g mol-1)
Heptanes	2.22	2.52	0.7324	91.3
Octanes	2.04	2.71	0.7433	106.6
Nonanes	1.53	2.30	0.7610	120.7
Decanes	1.27	2.24	0.7383	141.2
Calculated residue properties				
Heptanes plus				
Mole %		21.30		
Molecular Weight (g mol-1)		287		
Density at 15.6°C (g cm-3)		0.9072		
Decanes plus				
Mole %		15.51		
Molecular Weight (g mol-1)		355		
Density at 15.6°C (g cm-3)		0.9294		
Eicosanes plus				
Mole %		6.77		
Molecular Weight (g mol-1)		570		
Density at 15.6°C (g cm-3)		0.9885		
Hexatriacontanes plus				
Mole %		2.32		
Molecular Weight (g mol-1)		968		
Density at 15.6°C (g cm-3)		1.0736		
Whole sample properties				
Average mole weight (g mol-1)		80.2		



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CONSTANT COMPOSITION EXPANSION AT 110°C 4th GAS ADDITION - 1.024 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)		Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)
552.6		0.9959	1.3516		0.6419
545.7		0.9976	1.3539		0.6408
538.8		0.9994	1.3563		0.6396
536.3	Bubble point pressure	1.0000	1.3572	100.00	0.6392
532.3		1.0012	1.3588	89.85	
529.4		1.0021	1.3600	86.67	
525.7		1.0033	1.3616	84.60	
522.9		1.0041	1.3628	83.61	
519.8		1.0052	1.3642	82.76	
514.1		1.0070	1.3667	81.56	
488.0		1.0166	1.3797	77.91	
445.1		1.0358	1.4058	75.13	
379.3		1.0780	1.4631		
316.0		1.1429	1.5511		
263.2		1.2304	1.6699		
222.8		1.3348	1.8116		
189.4		1.4654	1.9888		
165.0		1.6031	2.1757		
144.6		1.7626	2.3921		
132.1		1.8896	2.5645		
103.2		2.3271	3.1582		
79.4		2.9747	4.0372		
68.6		3.4424	4.6719		
57.6		4.1255	5.5990		

Note.

Approximately 6.5 cm-3 of heavy liquid deposits, accumulated below bubble point pressure. The liquid volume % includes these liquid deposits.

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
- (2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
- (3) Liquid shrinkage as a percentage of volume at bubble point pressure.



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CALCULATED COMPOSITION OF MIXTURE FROM THE 4th HYDROCARBON GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	2.34	1.46
Nitrogen	0.86	0.34
Methane	61.21	13.96
Ethane	7.06	3.02
Propane	5.10	3.20
i-Butane	0.61	0.50
n-Butane	2.06	1.70
neo-Pentane	0.00	0.00
i-Pentane	0.67	0.68
n-Pentane	1.03	1.05
Hexanes	1.25	1.53
Me-Cyclo-pentane	0.37	0.44
Benzene	0.31	0.34
Cyclo-hexane	0.26	0.31
Heptanes	0.96	1.37
Me-Cyclo-hexane	0.30	0.42
Toluene	0.38	0.49
Octanes	1.05	1.70
Ethyl-benzene	0.18	0.27
Meta/Para-xylene	0.16	0.24
Ortho-xylene	0.10	0.14
Nonanes	0.85	1.54
Tri-Me-benzene	0.05	0.09
Decanes	1.02	2.05
Undecanes	0.96	2.00
Dodecanes	0.85	1.93
Tridecanes	0.84	2.09
Tetradecanes	0.72	1.94
Pentadecanes	0.72	2.11
Hexadecanes	0.58	1.81
Heptadecanes	0.53	1.79
Octadecanes	0.52	1.86
Nonadecanes	0.53	1.96
Eicosanes	0.38	1.47
Heneicosanes	0.34	1.41
Docosanes	0.32	1.39
Tricosanes	0.29	1.29
Tetracosanes	0.27	1.27
Pentacosanes	0.25	1.20
Hexacosanes	0.24	1.23
Heptacosanes	0.22	1.14
Octacosanes	0.21	1.16
Nonacosanes	0.20	1.11
Triacotanes	0.20	1.18

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**COMPOSITION RESIDUE PROPERTIES
 FROM THE 4th HYDROCARBON GAS ADDITION**

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.20	1.19		
Dotriacontanes	0.17	1.04		
Tritriacontanes	0.16	1.04		
Tetratriacontanes	0.15	1.01		
Pentatriacontanes	0.15	1.00		
Hexatriacontanes plus	1.93	26.55		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm ⁻³)	Molecular weight (g mol ⁻¹)
Heptanes	1.89	2.45	0.7326	91.3
Octanes	1.72	2.61	0.7434	106.6
Nonanes	1.28	2.20	0.7608	120.8
Decanes	1.07	2.14	0.7383	141.3
Calculated residue properties				
Heptanes plus				
Mole %		17.83		
Molecular Weight (g mol ⁻¹)		286		
Density at 15.6°C (g cm ⁻³)		0.9069		
Decanes plus				
Mole %		12.94		
Molecular Weight (g mol ⁻¹)		355		
Density at 15.6°C (g cm ⁻³)		0.9294		
Eicosanes plus				
Mole %		5.64		
Molecular Weight (g mol ⁻¹)		570		
Density at 15.6°C (g cm ⁻³)		0.9885		
Hexatriacontanes plus				
Mole %		1.93		
Molecular Weight (g mol ⁻¹)		968		
Density at 15.6°C (g cm ⁻³)		1.0736		
Whole sample properties				
Average mole weight (g mol ⁻¹)		70.4		

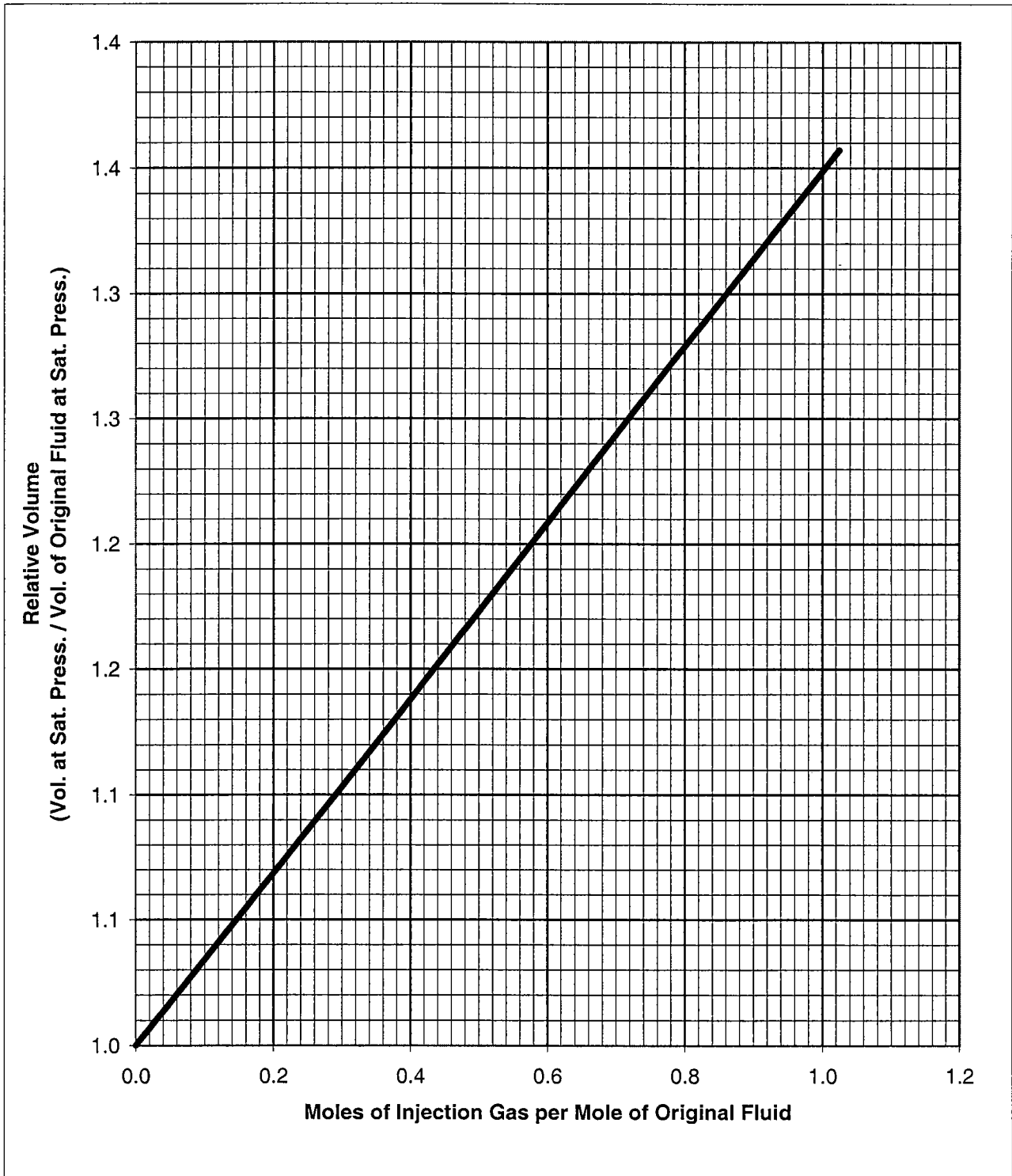


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RELATIVE VOLUME v MOLES OF HYDROCARBON INJECTION GAS FROM THE SOLUBILITY-SWELLING TEST AT 110°C



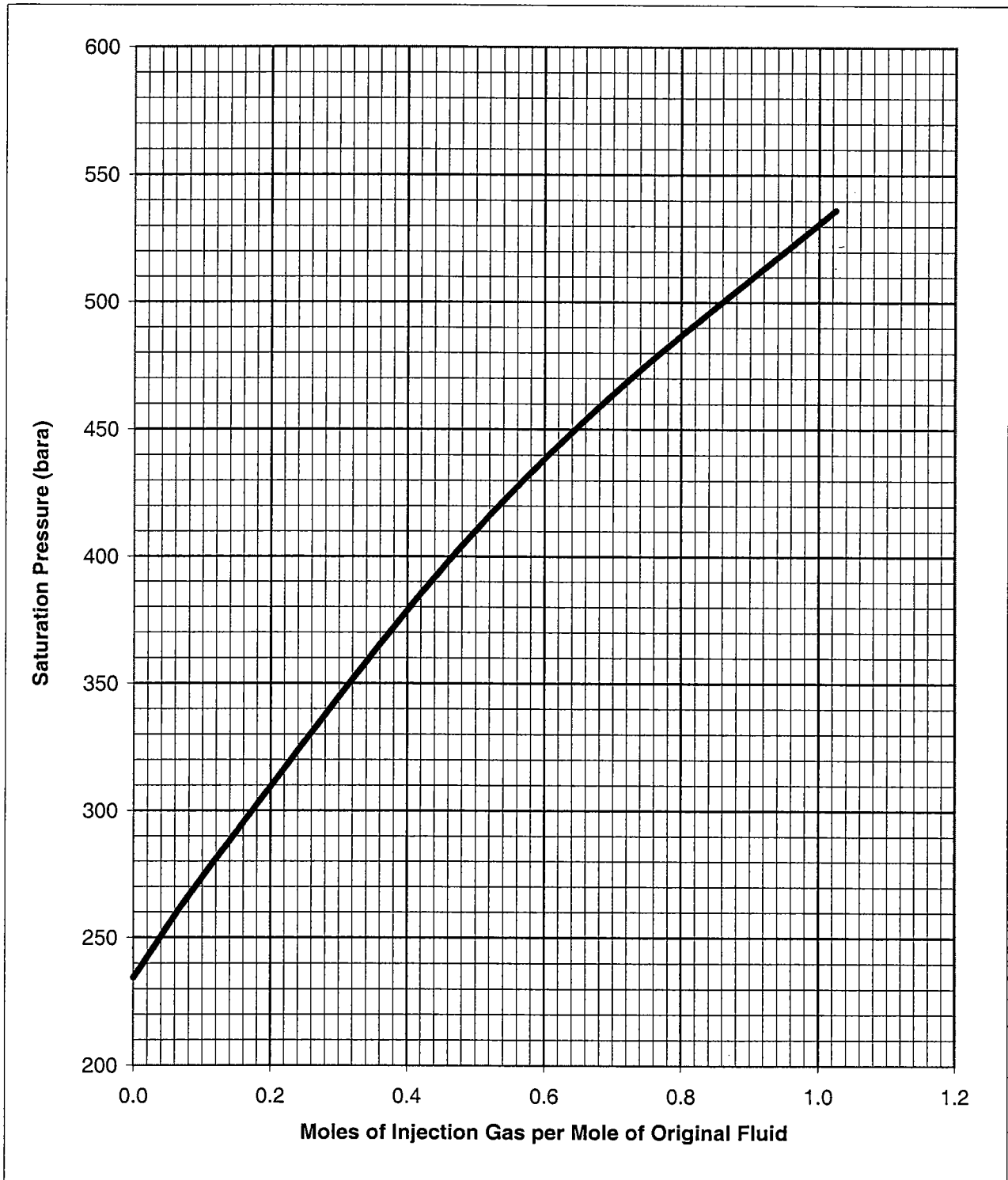


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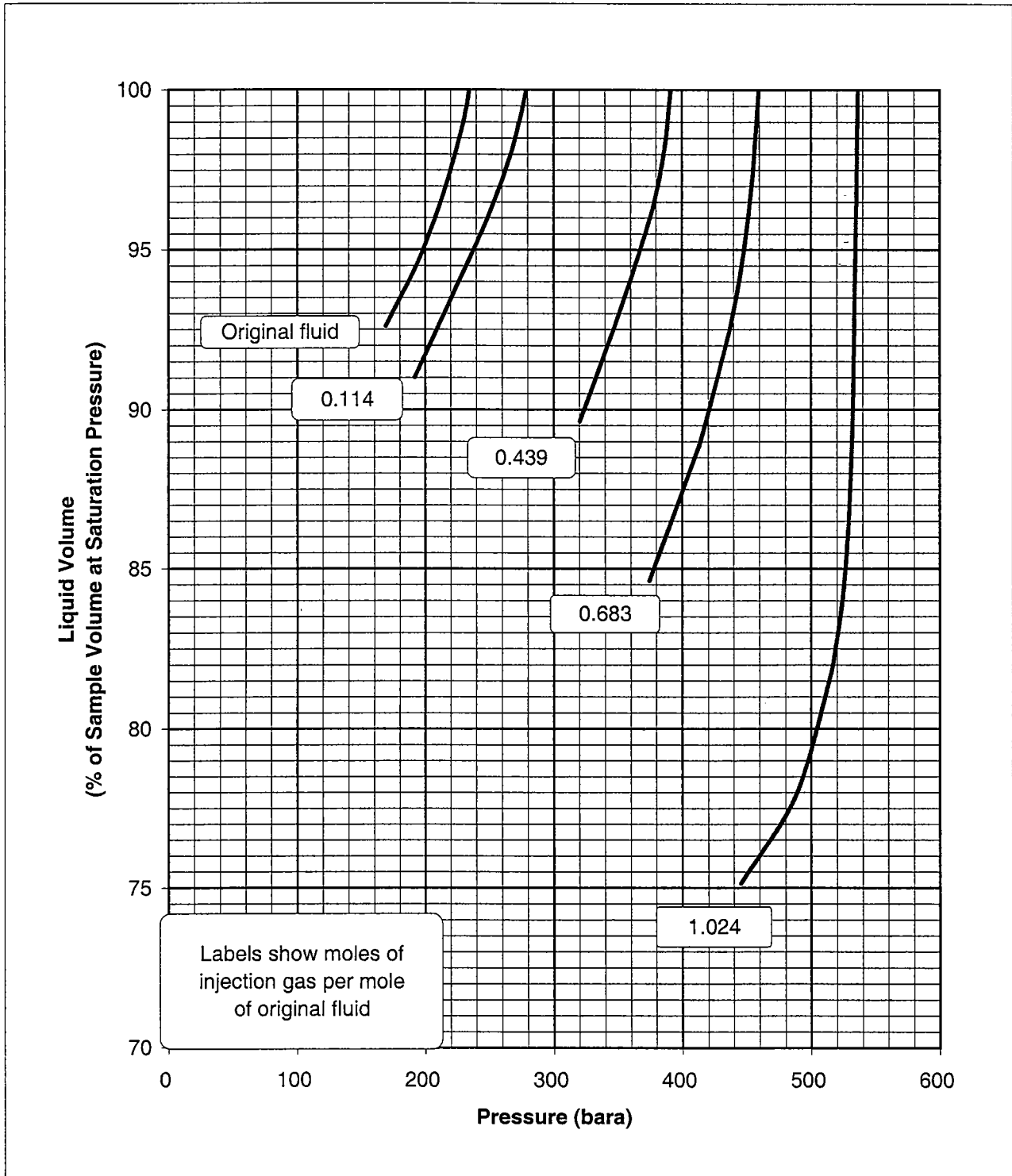
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SATURATION PRESSURE v MOLES OF HYDROCARBON INJECTION GAS FROM THE SOLUBILITY-SWELLING TEST AT 110°C



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**LIQUID VOLUME% v PRESSURE
FROM THE H-C GAS SOLUBILITY-SWELLING TESTS AT 110°C**





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Section D
Carbon Dioxide Gas Solubility Swelling



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CONSTANT COMPOSITION EXPANSION AT 110°C RECOMBINED SAMPLE

Pressure (bara)		Relative Volume (1)	Liquid Volume % (2)	Density (g cm ⁻³)	Y-function (3)
552.6		0.9570		0.7680	
414.7		0.9727		0.7557	
345.8		0.9820		0.7485	
276.8		0.9926		0.7405	
269.9		0.9938		0.7396	
263.0		0.9949		0.7387	
256.1		0.9961		0.7379	
249.2		0.9973		0.7370	
242.3		0.9986		0.7361	
235.4		0.9998		0.7351	
234.3	Bubble point pressure	1.0000	100.00	0.7350	
233.7		1.0006	99.82		
232.5		1.0020	99.51		
231.4		1.0031	99.29		
230.5		1.0041	99.10		
229.4		1.0053	98.91		
228.4		1.0064	98.74		
216.0		1.0213	97.01		
196.3		1.0506	94.81		3.831
168.8		1.1074	92.60		3.619
140.4		1.1971			3.394
117.0		1.3133			3.201
97.5		1.4624			3.036
81.8		1.6428			2.900
68.7		1.8677			2.781
59.2		2.0980			2.694
51.6		2.3499			2.622
39.7		2.9577			2.505
30.2		3.8025			2.408
21.8		5.2175			2.316

(1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.

(2) Liquid shrinkage as a percentage of volume at bubble point pressure.

(3) Y-function = $(P_{sat} - P) / (P_{abs})(V/V_{sat} - 1)$



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CONSTANT COMPOSITION EXPANSION AT 110°C 1st CO₂ GAS ADDITION - 0.114 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)		Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)
552.6		0.9563	0.9958		0.7689
483.6		0.9648	1.0047		0.7622
414.7		0.9741	1.0143		0.7549
345.8		0.9845	1.0251		0.7469
290.6		0.9938	1.0348		0.7399
283.7		0.9950	1.0361		0.7390
276.8		0.9963	1.0374		0.7381
269.9		0.9976	1.0388		0.7371
263.0		0.9989	1.0401		0.7362
257.2	Bubble point pressure	1.0000	1.0413	100.00	0.7353
257.0		1.0002	1.0415	99.93	
254.2		1.0028	1.0442	99.07	
253.2		1.0037	1.0451	98.80	
252.5		1.0044	1.0459	98.55	
244.5		1.0124	1.0542	96.85	
234.1		1.0241	1.0664	95.27	
216.7		1.0474	1.0907	93.44	
189.6		1.0965	1.1418	91.17	
156.2		1.1908	1.2400	88.40	
134.6		1.2860	1.3391		
107.1		1.4783	1.5394		
86.8		1.7175	1.7884		
63.3		2.2199	2.3115		
46.9		2.9025	3.0223		
32.7		4.0946	4.2637		

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
 (2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
 (3) Liquid shrinkage as a percentage of volume at bubble point pressure.



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CALCULATED COMPOSITION OF MIXTURE FROM THE 1st CARBON DIOXIDE GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	11.57	4.54
Nitrogen	0.50	0.13
Methane	37.69	5.39
Ethane	5.63	1.51
Propane	5.13	2.02
i-Butane	0.72	0.37
n-Butane	2.62	1.36
neo-Pentane	0.00	0.00
i-Pentane	0.95	0.61
n-Pentane	1.52	0.98
Hexanes	2.00	1.54
Me-Cyclo-pentane	0.60	0.45
Benzene	0.52	0.36
Cyclo-hexane	0.41	0.31
Heptanes	1.63	1.46
Me-Cyclo-hexane	0.51	0.45
Toluene	0.65	0.53
Octanes	1.83	1.86
Ethyl-benzene	0.32	0.31
Meta/Para-xylene	0.29	0.27
Ortho-xylene	0.17	0.16
Nonanes	1.49	1.71
Tri-Me-benzene	0.09	0.10
Decanes	1.80	2.28
Undecanes	1.70	2.23
Dodecanes	1.51	2.17
Tridecanes	1.51	2.36
Tetradecanes	1.30	2.20
Pentadecanes	1.30	2.38
Hexadecanes	1.03	2.05
Heptadecanes	0.95	2.02
Octadecanes	0.94	2.10
Nonadecanes	0.94	2.22
Eicosanes	0.67	1.66
Heneicosanes	0.61	1.59
Docosanes	0.58	1.57
Tricosanes	0.51	1.46
Tetracosanes	0.49	1.44
Pentacosanes	0.44	1.36
Hexacosanes	0.43	1.38
Heptacosanes	0.39	1.29
Octacosanes	0.38	1.31
Nonacosanes	0.35	1.26
Triacotanes	0.36	1.34

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**COMPOSITION RESIDUE PROPERTIES
FROM THE 1st CARBON DIOXIDE GAS ADDITION**

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.35	1.35		
Dotriacontanes	0.30	1.18		
Trtriacontanes	0.29	1.18		
Tetratriacontanes	0.27	1.14		
Pentatriacontanes	0.26	1.13		
Hexatriacontanes plus	3.47	29.99		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm-3)	Molecular weight (g mol-1)
Heptanes	3.17	2.58	0.7321	91.4
Octanes	2.99	2.84	0.7430	106.7
Nonanes	2.28	2.45	0.7614	120.7
Decanes	1.89	2.38	0.7384	141.2
Calculated residue properties				
Heptanes plus				
Mole %		31.66		
Molecular Weight (g mol-1)		289		
Density at 15.6°C (g cm-3)		0.9078		
Decanes plus				
Mole %		23.23		
Molecular Weight (g mol-1)		356		
Density at 15.6°C (g cm-3)		0.9296		
Eicosanes plus				
Mole %		10.15		
Molecular Weight (g mol-1)		570		
Density at 15.6°C (g cm-3)		0.9885		
Hexatriacontanes plus				
Mole %		3.47		
Molecular Weight (g mol-1)		968		
Density at 15.6°C (g cm-3)		1.0736		
Whole sample properties				
Average mole weight (g mol-1)		112.1		



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CONSTANT COMPOSITION EXPANSION AT 110°C 2nd CO₂ GAS ADDITION - 0.438 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)		Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm ⁻³)
552.6		0.9568	1.1049		0.7725
483.6		0.9667	1.1164		0.7645
414.7		0.9782	1.1297		0.7555
345.8		0.9922	1.1459		0.7449
338.9		0.9938	1.1477		0.7437
332.0		0.9955	1.1496		0.7425
325.1		0.9972	1.1516		0.7412
318.2		0.9989	1.1536		0.7399
314.0	Bubble point pressure	1.0000	1.1549	100.00	0.7391
312.7		1.0008	1.1558	99.09	
310.2		1.0024	1.1577	97.92	
308.0		1.0039	1.1593	97.22	
306.7		1.0047	1.1603	96.88	
302.9		1.0074	1.1634	96.03	
300.7		1.0089	1.1652	95.59	
299.8		1.0096	1.1659	95.41	
300.9		1.0087	1.1649	95.64	
296.9		1.0117	1.1684	94.86	
282.2		1.0239	1.1825	92.40	
261.0		1.0458	1.2077	89.58	
231.9		1.0874	1.2558		
193.9		1.1736	1.3553		
164.1		1.2838	1.4826		
141.0		1.4148	1.6339		
122.6		1.5658	1.8083		
106.1		1.7573	2.0294		
92.2		1.9806	2.2873		
73.7		2.4299	2.8062		
58.2		3.0474	3.5193		
43.1		4.0927	4.7264		

(1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.

(2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.

(3) Liquid shrinkage as a percentage of volume at bubble point pressure.



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CALCULATED COMPOSITION OF MIXTURE FROM THE 2nd CARBON DIOXIDE GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	31.21	14.16
Nitrogen	0.39	0.11
Methane	29.32	4.85
Ethane	4.38	1.36
Propane	3.99	1.81
i-Butane	0.56	0.34
n-Butane	2.04	1.22
neo-Pentane	0.00	0.00
i-Pentane	0.74	0.55
n-Pentane	1.18	0.88
Hexanes	1.55	1.38
Me-Cyclo-pentane	0.47	0.41
Benzene	0.41	0.33
Cyclo-hexane	0.32	0.28
Heptanes	1.27	1.31
Me-Cyclo-hexane	0.40	0.40
Toluene	0.50	0.48
Octanes	1.42	1.67
Ethyl-benzene	0.25	0.28
Meta/Para-xylene	0.22	0.25
Ortho-xylene	0.13	0.15
Nonanes	1.16	1.54
Tri-Me-benzene	0.07	0.09
Decanes	1.40	2.05
Undecanes	1.32	2.01
Dodecanes	1.18	1.95
Tridecanes	1.18	2.12
Tetradecanes	1.01	1.97
Pentadecanes	1.01	2.14
Hexadecanes	0.81	1.84
Heptadecanes	0.74	1.81
Octadecanes	0.73	1.88
Nonadecanes	0.74	1.99
Eicosanes	0.53	1.49
Heneicosanes	0.48	1.43
Docosanes	0.45	1.41
Tricosanes	0.40	1.31
Tetracosanes	0.38	1.29
Pentacosanes	0.34	1.22
Hexacosanes	0.34	1.24
Heptacosanes	0.30	1.16
Octacosanes	0.29	1.18
Nonacosanes	0.27	1.13
Triacotanes	0.28	1.20

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**COMPOSITION RESIDUE PROPERTIES
FROM THE 2nd CARBON DIOXIDE GAS ADDITION**

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.27	1.21		
Dotriacontanes	0.23	1.06		
Trtriacontanes	0.22	1.06		
Tetratriacontanes	0.21	1.02		
Pentatriacontanes	0.20	1.02		
Hexatriacontanes plus	2.70	26.96		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm-3)	Molecular weight (g mol-1)
Heptanes	2.46	2.33	0.7323	91.4
Octanes	2.32	2.55	0.7432	106.7
Nonanes	1.77	2.22	0.7611	120.7
Decanes	1.47	2.14	0.7383	141.2
Calculated residue properties				
Heptanes plus				
Mole %		24.63		
Molecular Weight (g mol-1)		289		
Density at 15.6°C (g cm-3)		0.9078		
Decanes plus				
Mole %		18.07		
Molecular Weight (g mol-1)		356		
Density at 15.6°C (g cm-3)		0.9296		
Eicosanes plus				
Mole %		7.90		
Molecular Weight (g mol-1)		570		
Density at 15.6°C (g cm-3)		0.9885		
Hexatriacontanes plus				
Mole %		2.70		
Molecular Weight (g mol-1)		968		
Density at 15.6°C (g cm-3)		1.0736		
Whole sample properties				
Average mole weight (g mol-1)		97.0		



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CONSTANT COMPOSITION EXPANSION AT 110°C
3rd CO₂ GAS ADDITION - 1.032 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)		Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)
552.6		0.9653	1.3080		0.7752
518.1		0.9717	1.3167		0.7701
483.6		0.9787	1.3262		0.7646
449.2		0.9866	1.3370		0.7584
435.4		0.9902	1.3417		0.7557
428.5		0.9920	1.3443		0.7543
421.6		0.9940	1.3469		0.7528
414.7		0.9960	1.3496		0.7513
407.8		0.9981	1.3525		0.7497
402.3	Bubble point pressure	1.0000	1.3551	100.00	0.7483
399.3		1.0011	1.3565	91.63	
397.5		1.0017	1.3573	88.94	
395.1		1.0026	1.3585	86.47	
392.7		1.0035	1.3598	84.83	
379.0		1.0090	1.3673	80.27	
360.9		1.0176	1.3789	77.12	
332.3		1.0350	1.4025	74.57	
290.2		1.0728	1.4537		
241.0		1.1484	1.5562		
206.3		1.2396	1.6798		
178.2		1.3553	1.8366		
159.2		1.4690	1.9907		
147.7		1.5583	2.1117		
136.4		1.6657	2.2572		
110.0		2.0381	2.7618		
87.5		2.5936	3.5145		
65.3		3.6069	4.8876		

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
- (2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
- (3) Liquid shrinkage as a percentage of volume at bubble point pressure.



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CALCULATED COMPOSITION OF MIXTURE FROM THE 3rd CARBON DIOXIDE GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	50.87	27.36
Nitrogen	0.28	0.10
Methane	20.94	4.10
Ethane	3.13	1.15
Propane	2.85	1.54
i-Butane	0.40	0.28
n-Butane	1.46	1.03
neo-Pentane	0.00	0.00
i-Pentane	0.53	0.47
n-Pentane	0.85	0.74
Hexanes	1.11	1.17
Me-Cyclo-pentane	0.34	0.34
Benzene	0.29	0.28
Cyclo-hexane	0.23	0.24
Heptanes	0.91	1.11
Me-Cyclo-hexane	0.29	0.34
Toluene	0.36	0.41
Octanes	1.02	1.42
Ethyl-benzene	0.18	0.23
Meta/Para-xylene	0.16	0.21
Ortho-xylene	0.10	0.12
Nonanes	0.83	1.30
Tri-Me-benzene	0.05	0.07
Decanes	1.00	1.74
Undecanes	0.95	1.70
Dodecanes	0.84	1.65
Tridecanes	0.84	1.80
Tetradecanes	0.72	1.67
Pentadecanes	0.72	1.81
Hexadecanes	0.58	1.56
Heptadecanes	0.53	1.53
Octadecanes	0.52	1.59
Nonadecanes	0.53	1.69
Eicosanes	0.38	1.26
Heneicosanes	0.34	1.21
Docosanes	0.32	1.19
Tricosanes	0.29	1.11
Tetracosanes	0.27	1.09
Pentacosanes	0.25	1.03
Hexacosanes	0.24	1.05
Heptacosanes	0.22	0.98
Octacosanes	0.21	1.00
Nonacosanes	0.20	0.96
Triacotanes	0.20	1.02



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COMPOSITION RESIDUE PROPERTIES FROM THE 3rd CARBON DIOXIDE GAS ADDITION

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.20	1.02		
Dotriacontanes	0.17	0.89		
Tritriacontanes	0.16	0.90		
Tetratriacontanes	0.15	0.86		
Pentatriacontanes	0.15	0.86		
Hexatriacontanes plus	1.93	22.82		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm ⁻³)	Molecular weight (g mol ⁻¹)
Heptanes	1.76	1.97	0.7321	91.4
Octanes	1.66	2.17	0.7430	106.7
Nonanes	1.27	1.86	0.7614	120.7
Decanes	1.05	1.81	0.7383	141.2
Calculated residue properties				
Heptanes plus				
Mole %		17.59		
Molecular Weight (g mol ⁻¹)		289		
Density at 15.6°C (g cm ⁻³)		0.9078		
Decanes plus				
Mole %		12.91		
Molecular Weight (g mol ⁻¹)		356		
Density at 15.6°C (g cm ⁻³)		0.9296		
Eicosanes plus				
Mole %		5.64		
Molecular Weight (g mol ⁻¹)		570		
Density at 15.6°C (g cm ⁻³)		0.9885		
Hexatriacontanes plus				
Mole %		1.93		
Molecular Weight (g mol ⁻¹)		968		
Density at 15.6°C (g cm ⁻³)		1.0736		
Whole sample properties				
Average mole weight (g mol ⁻¹)		81.9		



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CONSTANT COMPOSITION EXPANSION AT 110°C
4th CO₂ GAS ADDITION - 2.324 moles of GAS : 1.000 moles of SAMPLE

Pressure (bara)	Relative Volume (1)	Relative Volume (2)	Liquid Volume % (3)	Density (g cm-3)	
552.6	0.9953	1.7637		0.7731	
545.7	0.9972	1.7670		0.7716	
538.8	0.9991	1.7704		0.7701	
535.4	Dew point pressure	1.0000	1.7720	0.00	0.7694
528.2	1.0017	1.7750	19.13		
526.0	1.0022	1.7759	19.92		
523.5	1.0028	1.7770	20.33		
521.6	1.0033	1.7778	20.54		
519.3	1.0038	1.7788	20.76		
500.5	1.0089	1.7878	22.74		
478.3	1.0158	1.8001	25.70		
440.3	1.0307	1.8264	31.80		
382.0	1.0639	1.8852			
322.2	1.1199	1.9846			
271.7	1.1995	2.1256			
234.6	1.2923	2.2901			
206.9	1.3942	2.4705			
185.8	1.5010	2.6597			
169.8	1.6065	2.8468			
153.9	1.7407	3.0845			
131.3	2.0085	3.5591			
109.0	2.4145	4.2785			
85.4	3.1400	5.5641			

Retrograde liquid volumes quoted include a small volume of heavy liquid deposits that accumulated following CO₂ gas addition.

- (1) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume at bubble point pressure.
- (2) Relative Volume = V / V_{sat} ie. volume at indicated pressure per volume of original fluid at bubble point pressure.
- (3) Retrograde liquid accumulation as a percentage of volume at dew point pressure.



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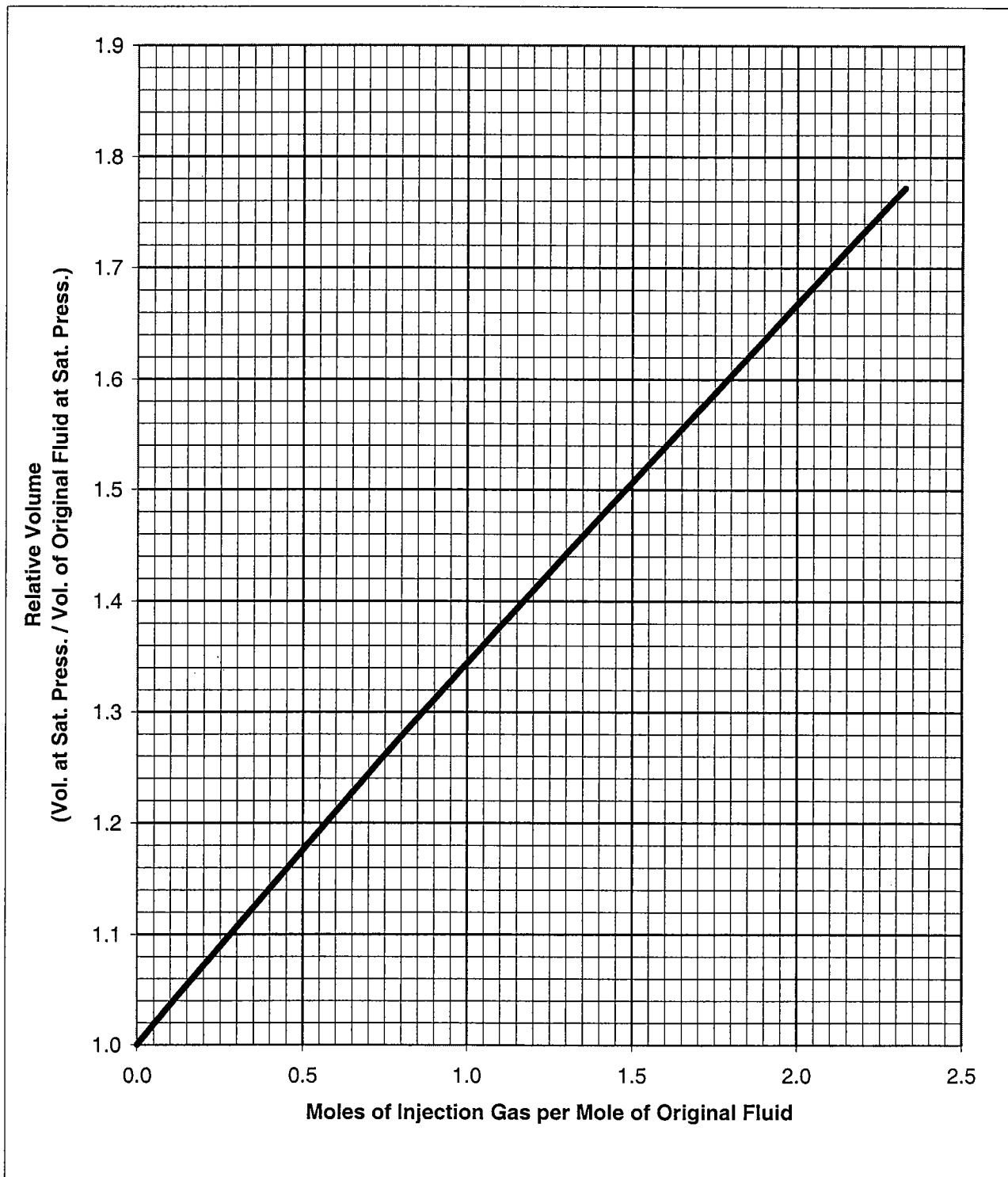
CALCULATED COMPOSITION OF MIXTURE FROM THE 4th CARBON DIOXIDE GAS ADDITION

Component	Mole %	Weight %
Hydrogen	0.00	0.00
Hydrogen sulphide	0.00	0.00
Carbon dioxide	70.44	46.45
Nitrogen	0.17	0.07
Methane	12.60	3.03
Ethane	1.88	0.85
Propane	1.71	1.13
i-Butane	0.24	0.21
n-Butane	0.88	0.76
neo-Pentane	0.00	0.00
i-Pentane	0.32	0.34
n-Pentane	0.51	0.55
Hexanes	0.67	0.86
Me-Cyclo-pentane	0.20	0.25
Benzene	0.17	0.20
Cyclo-hexane	0.14	0.17
Heptanes	0.54	0.82
Me-Cyclo-hexane	0.17	0.25
Toluene	0.22	0.30
Octanes	0.61	1.04
Ethyl-benzene	0.11	0.17
Meta/Para-xylene	0.10	0.15
Ortho-xylene	0.06	0.09
Nonanes	0.50	0.96
Tri-Me-benzene	0.03	0.05
Decanes	0.60	1.28
Undecanes	0.57	1.25
Dodecanes	0.51	1.22
Tridecanes	0.51	1.32
Tetradecanes	0.43	1.23
Pentadecanes	0.43	1.34
Hexadecanes	0.35	1.15
Heptadecanes	0.32	1.13
Octadecanes	0.31	1.18
Nonadecanes	0.32	1.24
Eicosanes	0.23	0.93
Heneicosanes	0.20	0.89
Docosanes	0.19	0.88
Tricosanes	0.17	0.82
Tetracosanes	0.16	0.81
Pentacosanes	0.15	0.76
Hexacosanes	0.14	0.78
Heptacosanes	0.13	0.72
Octacosanes	0.13	0.73
Nonacosanes	0.12	0.71
Triacotanes	0.12	0.75

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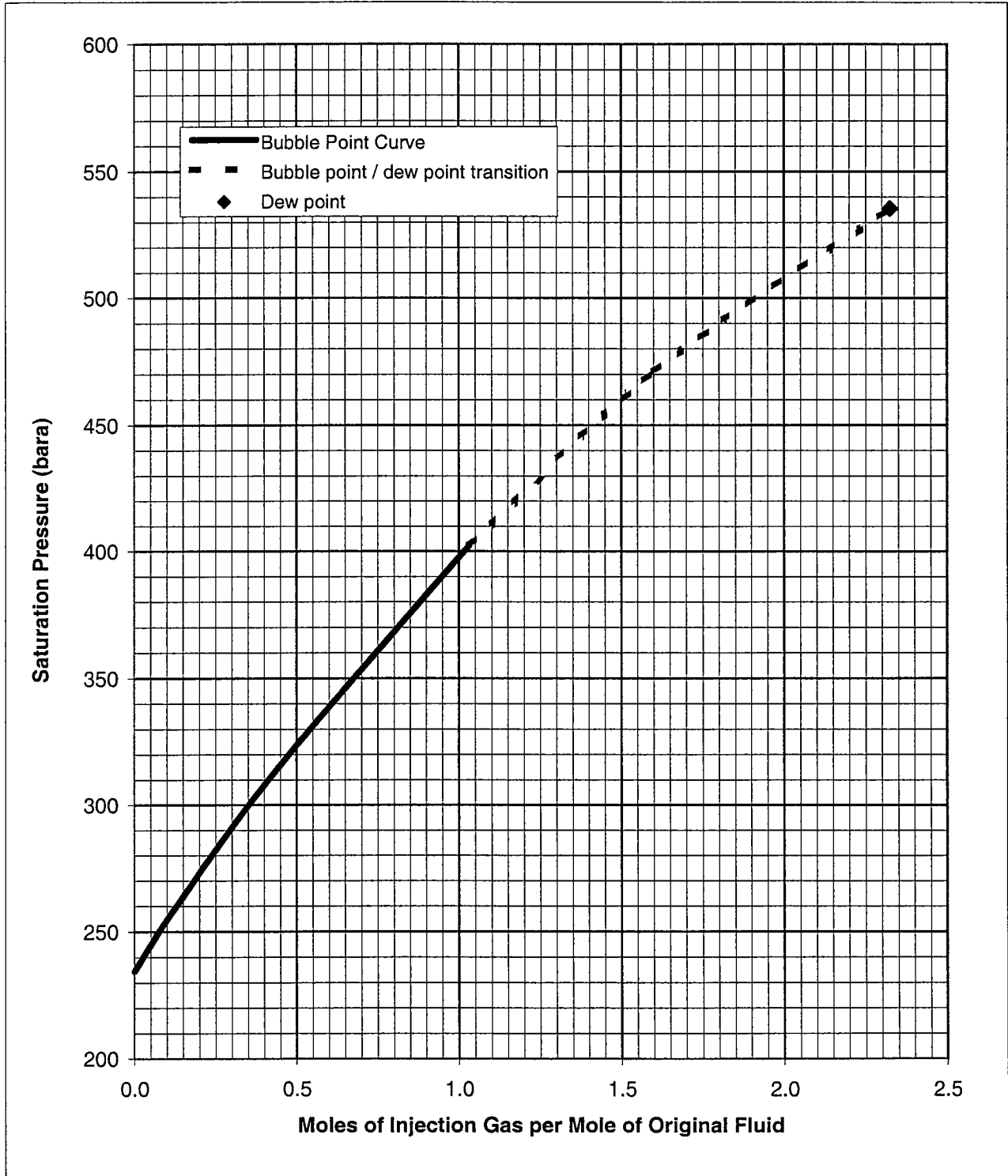
**COMPOSITION RESIDUE PROPERTIES
 FROM THE 4th CARBON DIOXIDE GAS ADDITION**

Components (continued)	Mole %	Weight %		
Hentriacontanes	0.12	0.76		
Dotriacontanes	0.10	0.66		
Trtriacontanes	0.10	0.66		
Tetratriacontanes	0.09	0.64		
Pentatriacontanes	0.09	0.63		
Hexatriacontanes plus	1.16	16.83		
Totals :	100.00	100.00		
Grouped Data	Mole %	Weight %	Density at 15.6°C (g cm-3)	Molecular weight (g mol-1)
Heptanes	1.06	1.44	0.7314	91.5
Octanes	1.00	1.59	0.7432	106.7
Nonanes	0.76	1.37	0.7609	120.7
Decanes	0.63	1.33	0.7380	141.3
Calculated residue properties				
Heptanes plus				
Mole %		10.58		
Molecular Weight (g mol-1)		289		
Density at 15.6°C (g cm-3)		0.9078		
Decanes plus				
Mole %		7.76		
Molecular Weight (g mol-1)		356		
Density at 15.6°C (g cm-3)		0.9296		
Eicosanes plus				
Mole %		3.39		
Molecular Weight (g mol-1)		570		
Density at 15.6°C (g cm-3)		0.9885		
Hexatriacontanes plus				
Mole %		1.16		
Molecular Weight (g mol-1)		968		
Density at 15.6°C (g cm-3)		1.0736		
Whole sample properties				
Average mole weight (g mol-1)		66.8		

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RFLA 980053**RELATIVE VOLUME v MOLES OF CO₂ INJECTION GAS
FROM THE SOLUBILITY-SWELLING TEST AT 110°C**

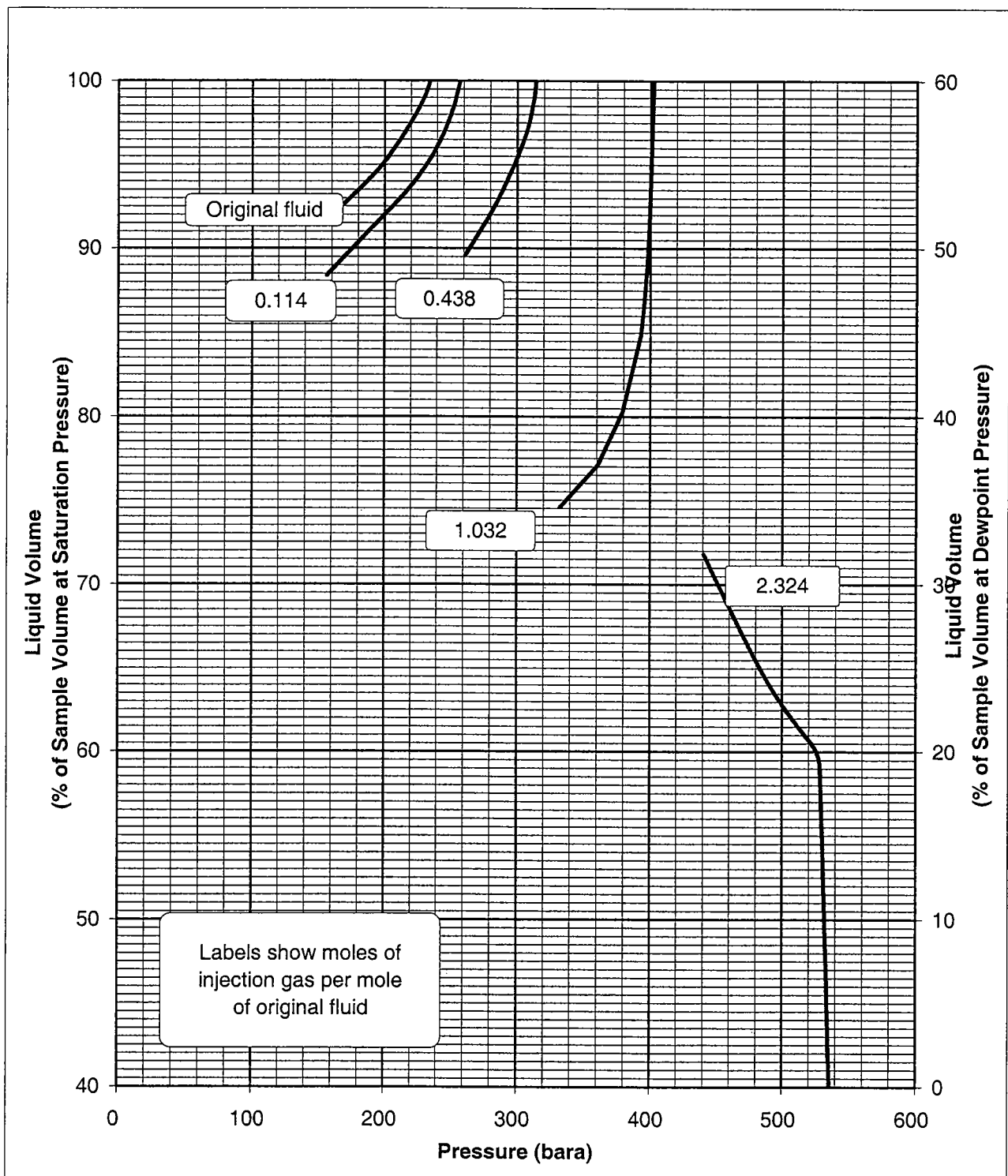
Statoil a.s.
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RFLA.980053

**SATURATION PRESSURE v MOLES OF CO₂ INJECTION GAS
FROM THE SOLUBILITY-SWELLING TEST AT 110°C**



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**LIQUID VOLUME% v PRESSURE
FROM THE CO₂ SOLUBILITY-SWELLING TESTS AT 110°C**





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Section E
Appendix



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DATA USED IN GAS COMPOSITIONAL CALCULATIONS

Component	Mole Weight (g mole ⁻¹)	Density (g cm ⁻³ at 15.6°C)	Component	Mole Weight (g mole ⁻¹)	Density (g cm ⁻³ at 15.6°C)
Hydrogen	* 2.016	N/A	33DMC5	* 100.20	0.6954
Oxygen/(Argon)	** 31.999	1.1410	Cyclohexane	* 84.16	0.7827
Nitrogen (Corrected)	** 28.013	0.8086	2MC6/23DMC5	* 100.20	0.6917
Methane	** 16.043	0.2997	11DMCYC5/3MC6	* 99.20	0.7253
Carbon Dioxide	** 44.010	0.8172	t13DMCYC5	* 98.19	0.7528
Ethane	** 30.070	0.3558	c13DMCYC5/3EC5	* 99.20	0.7262
Hydrogen Sulphide	** 34.080	0.8006	t12DMCYC5	* 98.19	0.7554
Propane	** 44.097	0.5065	Heptanes (nC7)	* 100.20	0.6875
i-Butane	** 58.123	0.5623	22DMC6	* 114.23	0.6994
n-Butane	** 58.123	0.5834	MCYC6	* 98.19	0.7740
Neo-Pentane	* 72.15	0.5968	ECYC5	* 98.19	0.7704
i-Pentane	** 72.150	0.6238	223TMC5/24&25DMC6	* 114.23	0.7060
n-Pentane	** 72.150	0.6305	ctc124TMCYC5	* 112.21	0.7511
22DMC4	* 86.18	0.6529	ctc123TMCYC5	* 112.21	0.7574
23DMC4/CYC5	* 78.16	0.7129	Toluene	* 92.14	0.8734
2MC5	* 86.18	0.6572	Octanes (nC8)	* 114.23	0.7063
3MC5	* 86.18	0.6682	E-Benzene	* 106.17	0.8735
Hexanes (nC6)	* 86.18	0.6631	M/P-Xylene	* 106.17	0.8671
22DMC5	* 100.20	0.6814	O-Xylene	* 106.17	0.8840
M-C-Pentane	* 84.16	0.7533	Nonanes (nC9)	* 128.26	0.7212
24DMC5	* 100.20	0.6757	Decanes	*** 134	0.778
223TMC4	* 100.20	0.6947	Undecanes	*** 147	0.789
Benzene	* 78.11	0.8820	Dodecanes	*** 161	0.800

Data Source Refs :

* ASTM Data Series Publication DS 4B (1991) - Physical Constants of Hydrocarbon and Non-Hydrocarbon Compounds

** GPA Table of Physical Constants of Paraffin Hydrocarbons and Other Components of Natural Gas
GPA 2145-96

*** Journal of Petroleum Technology, Nov 1978, Pages 1649-1655
Predicting Phase Behaviour of Condensate/Crude Oil Systems Using Methane Interaction Coefficients
- D.L. Katz & A. Firoozabadi

Note :

The gas mole % compositions were calculated from the measured weight % compositions using the most detailed analysis results, involving as many of the above components as were identified. The reported component mole % compositions were then sub-grouped into the generic carbon number components.



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DATA USED IN LIQUID COMPOSITIONAL CALCULATIONS

Component	Mole Weight (g mole ⁻¹)	Density (g cm ⁻³ at 15.6°C)	Component	Mole Weight (g mole ⁻¹)	Density (g cm ⁻³ at 15.6°C)
Hydrogen	* 2.016	N/A	Undecanes	*** 147	0.789
Hyd. sulphide	** 34.080	0.8006	Dodecanes	*** 161	0.800
Carbon Dioxide	** 44.010	0.8172	Tridecanes	*** 175	0.811
Nitrogen	** 28.013	0.8086	Tetradecanes	*** 190	0.822
Methane	** 16.043	0.2997	Pentadecanes	*** 206	0.832
Ethane	** 30.070	0.3558	Hexadecanes	*** 222	0.839
Propane	** 44.097	0.5065	Heptadecanes	*** 237	0.847
i-Butane	** 58.123	0.5623	Octadecanes	*** 251	0.852
n-Butane	** 58.123	0.5834	Nonadecanes	*** 263	0.857
i-Pentane	** 72.150	0.6238	Eicosanes	*** 275	0.862
n-Pentane	** 72.150	0.6305	Heneicosanes	*** 291	0.867
Hexanes	** 86.177	0.6634	Docosanes	*** 305	0.872
Me-cyclo-pentane	* 84.16	0.7533	Tricosanes	*** 318	0.877
Benzene	* 78.11	0.8820	Tetracosanes	*** 331	0.881
Cyclo-hexane	* 84.16	0.7827	Pentacosanes	*** 345	0.885
Heptanes	** 100.204	0.6874	Hexacosanes	*** 359	0.889
Me-cyclo-hexane	* 98.19	0.7740	Heptacosanes	*** 374	0.893
Toluene	* 92.14	0.8734	Octacosanes	*** 388	0.896
Octanes	** 114.231	0.7061	Nonacosanes	*** 402	0.899
Ethyl-benzene	* 106.17	0.8735	Triacontanes	*** 416	0.902
Meta/Para-xylene	* 106.17	0.8671	Hentriacontanes	*** 430	0.906
Ortho-xylene	* 106.17	0.8840	Dotriacontanes	*** 444	0.909
Nonanes	** 128.258	0.7212	Tritriacontanes	*** 458	0.912
1-2-4-T-M-benzene	* 120.19	0.8797	Tetracontanes	*** 472	0.914
Decanes	** 142.285	0.7334	Pentatriacontanes	*** 486	0.917
			Hexatriacontanes	*** 500	0.919

Data Source Refs :

* ASTM Data Series Publication DS 4B (1991) - Physical Constants of Hydrocarbon and Non-Hydrocarbon Compounds

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Note :

The residue mole weight and density values (eg heptanes plus, undecanes plus, eicosanes plus) are calculated so that the calculated average mole weights and densities correspond with the measured values. This can lead to anomalous residue mole weights and densities where the Katz and Firoozabadi values may not be suitable for the isomer groups detected.



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